

The potential for dust detection by means of µXRF scanning in Eifel maar lake sediments

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Abstract:

Data for the annual variability of aeolian sediments is obtained from continuous and high resolution µXRF geochemistry within maar lake sediments from the last 60 kyrs. Two sediment cores from Eifel maar lakes and dry maars (Germany) were analyzed, which covering the glacial inception of MIS-3, LGM and MIS-2, transition I as well as the Holocene. The energy dispersive XRF scanning is obtained on resin impregnated blocks of the sediment, which are the basis for the production of petrographic thin section. Thus, the measurement results can directly be compared with micro facies analysis. Quantification of the Eagle III µXRF carried out on one sediment core, using the standard free fundamental parameter method, shows that this quantification method gives suitable results by comparison with WD-XRF analysis of discrete samples. Each single maar differs in lithological composition, which is reflected in the geochemistry, too. The major processes of element deposition in lakes are therefore described, i.e. the different sedimentation and weathering processes as well as the circulation of the water body of the lake. Following on from this, it is shown that it is possible to derive an aeolian sediment signal by using principle component analysis of standardized variables. Because further knowledge about the lithology and environmental background is available (from the petrographic thin sections) it can be demonstrated that this principle component analysis approach gives reliable results for all the time slices investigated. The most prominent element for describing dust in both investigated cores Ca, which reaches highest values (>5 wt.-%) during glacial conditions and which have major influence on the dust factor obtained by principal component analysis. In combination with grayscale values both the Ca content and the dust factor serve to, calculated by principal component analysis, the serves to record aeolian dust in laminated lake sediments. In both cores periods with major dust events could be detected by µXRF geochemistry: during MIS-3 the largest Heinrich event H4 and the onset of dust increase coupled to the glacial inception of the Pleni-Weichselian, further on the whole MIS-2 including LGM and YD as well as enhanced dust supply forced by human activities since the Subboreal.

[Möglichkeiten der Detektion äolischen Staubs mittels µXRF-Scannen von Maarseesedimenten aus der Eifel]

Kurzfassung: Mittels kontinuierlicher und hochauflösender µXRF-Geochemieanalysen wird die Variabilität äolischer Sedimente der letzten 60.000 Jahre rekonstruiert. Dazu werden zwei Sedimentbohrkerne jeweils aus einem Maarsee und einem Trockenenmaar (Eifel. Deutschland) untersucht. Beide Kerne umfassen das letzte Glazial, einschließlich des MIS-3, des LGM und MIS-2, Transition I als auch das Holozän. Die energiedispersive RFA-Messungen der Eagle III µXRF wird direkt an Harz imprägnierten Proben angewendet. Diese sogenannten Tränklinge bilden die Grundlage für die Herstellung von petrographischen Dünnschliffen und somit können die Messergebnisse direkt mit einer Mikrofaziesanalyse verglichen werden. Anhand eines Sedimentkerns wird gezeigt, dass eine Quantifizierung der µXRF-Ergebnisse mittels der Fundamentalparametermethode geeignete ist. Eine Überprüfung der Ergebnisse findet dabei mit wellenlängen-dispersiven RFA-Messungen an diskreten Proben statt. Die Ergebnisse zeigen, dass sich jedes einzelne Maar in der lithologischen Zusammensetzung und damit auch geochemisch unterscheidet. Deshalb wird auf die grundlegenden Prozesse der Elementdeposition in die Seen eingegangen, die mit der Ablagerung von Sedimenten, der Variabilität der chemischen Verwitterung oder der Wasserzirkulation in Zusammenhang stehen. Mittels Hauptkomponentenanalysen standardisierter Variablen ist darüber hinaus die objektive Ableitung eines äolischen Sedimentsignals möglich. Es wird gezeigt, dass dieser Ansatz verlässliche Ergebnisse für alle untersuchten Zeitabschnitte liefert, solange für die Interpretation weitere Kenntnisse über die Lithologie und Paläoökologie zur Verfügung stehen. Das auffälligste Element zur Charakterisierung von Staub ist in beiden untersuchten Kernen Kalzium. Die höchsten Werte (>5 Gew.-%) werden während vollglazialer Bedingungen erreicht. Kalzium hat einen wesentlichen Einfluss auf den Staubfaktor der Hauptkomponentenanalyse. Eine zusätzliche Kombination der Kalziumgehalte mit dem Staubfaktor der Hauptkomponentenanalyse sowie Grauwertmessungen verbessert den Nachweis äolischen Staubs in laminierten Seesedimenten zusätzlich. In beiden Kernen konnten Sedimente mit erhöhten Staubkonzentrationen geochemisch nachgewiesen werden: Während des MIS-3 sind das vor allem das größte Heinrich-Ereignis H4 sowie der Anstieg des atmosphärischen Staubgehalts während der Wiedervereisung der Inlandsgletscher. Weiterhin ist das gesamte MIS-2 einschließlich LGM und der Jüngeren Dryas von starker Staubdeposition charakterisiert. Eine erhöhte Staubkonzentration ist ebenfalls ab dem Subboreal nachgewiesen und wird als anthropogene Aktivität gedeutet.

Keywords:

 μ XFR, geochemistry, lacustrine sediment, aeolian sediment, last glacial cycle, PCA

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1 Introduction

The investigation of long continuous aeolian sediment records, such as loess profiles or dunes are widely used to reconstruct the environmental and climate variability. Long records cover several glacial cycles and are primarily known from the large loess plains in Asia or from the Carpathian basin. Profiles with the possibility to archive annual resolution are less common. The West-Eifel Volcanic Field (WEVF), Germany, provides such records in maar lakes. Without a fluvial inlet, these lakes are perfect sediment traps for aeolian deposits (DIETRICH & SIROCKO 2009, PFAHL et al. 2009).



Fig. 1: Geological sketch map of the Eifel region and position of the cores SM3 (maar lake Schalkenmehren) and DE3 (Dehner dry maar). The map is based on the Geological Map of Germany (BGR, 1993), locations of dry maars after BÜCHEL (1994).

Abb. 1: Geologische Übersichtskarte der Eifel mit den Positionen der beiden Bohrkerne SM3 (Schalkenmehrener Maar) und DE3 (Dehner Trockenmaar). Die Karte basiert auf der Geologischen Übersichtskarte von Deutschland (1:1 000 000; BGR, 1993), Lokationen der Trockenmaare nach BÜCHEL (1994).

Furthermore, although the sediments do not show annual lamination, they are stratified by event layers due to anoxic conditions at the lake bottom (ZOLITSCHKA et al. 2000, SIROCKO et al. 2005).

The detection of aeolian deposits in lake sediments is possible by using different approaches like micro facies analysis, measuring the minerogenous component of the bulk sediment (e.g. ZOLITSCHKA 1998), magnetic susceptibility (e.g. YANCHEVA et al. 2007) or grain size analysis (e.g. SEELOS et al. 2009, SUN et al. 2002). However, in this study a geochemical scanning approach is applied to measure aeolian deposits. High resolution geochemical analysis of lacustrine sediments offers a reliable method for investigation of environmental changes in the past (BOYLE 2001, BOYLE 2000, ENGSTROM & WRIGHT 1984, MELLES et al. 2007). The potential of X-Ray fluorescence (XRF) core scanning is based on the rapid and non-destructive acquisition of high-resolution geochemical data from lacustrine sediment cores. This facilitates new approaches to many applications in paleolimnology, including pollution detection, varve counting, and estimation of past ecosystem productivity (FRANCUS et al. 2009). The μ XRF scanner acquires bulk-sediment chemical data with sufficient accuracy for major elements (BOYLE 2000). Although elemental intensities are predominantly proportional to concentration, they are also influenced by the energy level of the

X-ray source, the count time, and the physical properties of the sediment, such as the poorly constrained measurement geometry attributable to inhomogeneity of the specimens (e.g. variable water content and grain-size distribution), and irregularities of the sample surface (Weltje & TJALLINGII 2008, ROTHWELL et al. 2006). Nevertheless, the XRF methodology is a widely accepted, semi-quantitative core logging method that provides records of changing element intensities expressed in "total counts", reflecting the geochemical composition of the sediments. Quantitative analysis is made more difficult by matrix effects especially for light elements such as Al and Si (BÖNING et al. 2007). Reasons include the pore volume of interstadial water or resin or the roughness of the sediment surface. However, a recent approach to calibration of XRF core scanners for quantitative geochemical logging is applied by WELTJE & TJALLINGII (2008) or KIDO et al. (2006). Beside the direct measurement of the core surface via core loggers the analysis of resin impregnated samples (RIS) is demonstrated in a growing number of studies. The single 10 cm long RIS are taken continuously down core. The pore volume of the sediment is substituted by resin during the production of the RIS. These geochemical data can be directly assigned to micro-facies data because XRF scanning has been carried out on the same impregnated sediment blocks from which thin sections can be prepared. High resolution applications are shown by different studies for example by SORREL et al. (2007) or BRAUER et al. (2008). The latter authors have applied a geo-chemical major element micro-X-ray fluorescence scanner at 50 µm resolution providing geochemical information for individual seasonal layers (5-8 data points/ varve for the Allerød; 20-30 data points/varve for Younger Dryas) to investigate the structure and seasonal composition of varves. The first successful implementation of XRF techniques for the provenance analysis of dust is shown by NEFF et al. (2008).

The objective of the work is to separate of different lithofacies, especially the aeolian input into the lake environment, by means of μ XRF geochemistry and to test if μ XRF data are not only a general paleoclimate signal but can be used to quantify the aeolian fraction directly to achieve dust records with annual resolution. In this study, we demonstrate how aeolian sediment within maar lake sediments might be classified by certain elements and new factors, calculated by principal component analysis (PCA).

2 Material and Methods

2.1 Cores and drilling locations

The west Eifel volcanic field (Germany) is characterized by more than 70 maar lakes and dry maar lakes (BÜCHEL 1994). Two sediment cores, namely SM3 (N50°10'11.6", E6°51'31"), from lake Schalkenmehren and DE3 (N50°17'35.5", E6°30'22.7", 88 m depth) from the Dehner dry maar (Dehner Trockenmaar) were drilled in a maar lake and a dry maar, respectively (Fig.1). On the Eastern side of lake Schalkenmehrener Maar an older dry maar is located which is connected to the lake by a marsh. Earlier fluvial input into the maar lake from this extended catchment could be proved until the Medieval Times (STRAKA 1975). The rim of the Dehner dry maar exhibits a pronounced roundness and shows no ancient fluvial inflow but does have an outflow to the Northwest. The core SM3 from lake Schalkenmehrener Maar was drilled using a swimming platform and a piston corer (UWITEC, Austria), DE3 was drilled with the 'Seilkernverfahren'. These two cores from the ELSA repository (Eifel Laminated Sediment Archive) cover together the main periods with major climate changes of the last 60 kyrs, including the marine isotope stages MIS-3 and MIS-2 (DE3), and transition I as well as (anthropogenic) dust events during the Holocene (SM3). In total 82 meters of sediment cores with an average sedimentation rate of 1.5±0.5 mm/yr are analyzed. The sediments consist of detrital bearing sediment that documents weather extremes (flash floods and aeolian dust) or gyttja, containing a high amount of organic matter that documents past water conditions (temperature, nutrient content, pH), which are recorded in the remains of plants and animals that lived in the lake.

2.2 Chronology and lithology

The stratigraphy for both cores is published in SIROCKO (Ed. 2009). The core SM3 spans the Holocene and the core DE3 from the Dehner dry maar covers the time span from MIS 4 to the eruption of Lake Laach (Laacher See). In DE3 a piece of spruce could be dated by radiocarbon to 45.8 + 1.2 - 1.04 kyrs BP at 55.56 m depth. Between these time marks are four stadial-interstadial sequences shown by high organic matter and therefore dark brownish color (Fig. 2). The tephra of the eruption of lake Laach (the Lacher See Tephra, LST, at 12.9 kyrs BP, van den BOGAARD 1995, WÖRNER & SCHMINCKE 1984) is identified geochemically (VERES, pers. comm.) in the core, SM3 at 6.5 m and in DE3 at 3.4 m depth. Thus, the LST is used as a tie point to merge both cores to a stack, which covers the last 60 kyrs.

Core DE3 (Fig. 2) shows significant changes in environmental and climatic conditions in its sediments and oscillates between warm and organic-rich and dust-rich glacial conditions. Starting at the bottom sequence of the core, dark brown gyttja is dominant and implicates an interstadial character between 70 and 50 m. At 43 to 41 m depth single but rather strong dust events had been deposited and the sediment has a brighter color until 40.50-38.90 m where dark, organic-rich matter recurs. Volcanic ashes from Eifel volcanism, mostly with unknown certain origin, cumulates between 80-78 m, and between 51-37 m. In the core meter 36 a slide is recorded as a decimeter large fold. The following ten meters are dominated by light brown gyttja with increasing dust influence. The content of aeolian sediment reaches the highest level at 26 m and keeps stable until 10 m depth. The next eight meters dust is decreasing and at 8 m the first grass pollen appears (SIкоско ed. 2009). Between 7 and 2.8 m the content on organic matter is increasing, whereas the uppermost core sequence is characterized by debris flow/solifluction debris. The core SM3 (Fig. 3) is fine laminated in the uppermost 6 meters with an high amount of organic gyttja between 3-5 m. From 6.5 m down core, the sediment changes significantly to loess gyttja with a bright grey color, showing a strong influence of aeolian sediments during this period.



Fig. 2: Core photography and time marks (tuned to the Greenland ice core GICC05 stratigraphy (SVENSSON et al. 2008) of the core DE3. The ash of the LST is clearly visible at 3.5 m depth. The core sections between 35 and 10 m is dominated by single strong dust events (SEELOS et al. 2009; DIETRICH & SEELOS 2010) and represent the time of the Middle Weichselian. Modified after SIROCKO (Ed. 2009).

Abb. 2: Kernfotografie und Altersmarken des Sedimentkerns DE3. Das Altersmodell basiert auf 14C-Datierungen, Tephrochronologie und ist mit der GICC05-Stratigrafie der Grönländischen Eiskerne abgestimmt (SVENSSON et al. 2008). Die Laacher-See-Tephra (LST) ist in 3,5 m Tiefe deutlich erkennbar. Die Kernsequenz zwischen 35 und 10 m wird durch äolischen Staub dominiert (SEELOS et al. 2009; DIETRICH & SEELOS 2010) und repräsentiert den Zeitraum der Weichsel-Vereisung. Überarbeitet nach SIROCKO (Ed. 2009).



Fig. 3: Core photography and time marks of the core SM3. The ash of the LST is clearly visible (6.3 m depth). The deeper core sequence is dominated by bright colored aeolian sediment of the late Weichselian.

Abb. 3: Kernfotografie und Altersmarken des Sedimentkerns SM3. Die LST ist in 6,3 m Tiefe deutlich zu sehen. Die untere Kernsequenz ist durch die hellen äolischen Sedimente der späten Weichselkaltzeit gekennzeichnet.

2.3 Evaluation of chemical data

The three core sections were scanned for selected elements at 1 mm resolution using the energy dispersive X-ray fluorescence scanner Eagle III (Röntgenanalytik Meßtechnik GmbH, Germany) of the University of Mainz (Germany). The µXRF was used to measure major elements (e.g., Mg, Al, Si, P, S, K, Ca, Ti, Mn, and Fe) as well as trace elements (Sr, Zr, Cr, Ni, Cu, and Zn) on 10 cm long resin impregnated samples. Na is too light for detection via the Eagle III and is therefore not included in this study. The Araldite(R) impregnated blocks are a spin-off from to the production of thin sections (DIETRICH & SIROCKO 2009). Thus, the geochemical results can be directly interpreted in context with micro facies analysis. The µXRF scanner acquires bulksediment chemical data from these samples. Scans of pure Araldite indicate zero counts of the investigated element and thus do not bias the results. Although elemental intensities are dominantly proportional to concentration, they are also influenced by the energy level of the X-ray source, the count time, and the physical properties of the sediment (Röhl & Abrams 2000). The data were collected every 1 mm down-core with a 300 µm spot size of the X-ray beam. The generator setting was of 30 kV and 500 μA with a sampling time of 16 seconds and a mean death time of 30 µs. The sample chamber is evacuated during measurement, thus light elements like Si and Al are also stimulated. The emitted fluorescence radiation is detected by an energy dispersive Si(Li)-detector with an area of 30 mm² and an energy resolution of 135 eV at maximum.

Discrete reference samples are taken from core DE3 and glass beads with a five times dilution are measured by a Philips MagiXPRO via WD-XRF method (Rh-anode, 3.2 KW). The quantification is based on international standards (Govindaraju 1989). This study also examines whether a quantification of the Eagle III μ XRF using the standard free fundamental parameter method is suitable. The alternative is using the intensities as relative results in combination with quantified but independent measurements by WD-XRF methodology.

2.4 Statistical processing of the data

Core sequences with reworked material were removed from the original data set. High amounts of resin (e.g. cracks caused by sample preparation) were detected by a locally low sum of measured intensities and are subsequently removed. Therefore a threshold of 500 cps was evaluated empirically.

A principal component analysis (PCA) was performed to detect different lithological facies, such as dust, by using inorganic geochemical data. The PCA was used to objectively describe the differences and similarities between the elements and to reduce the number of variables by showing the main variance of the data set by only a few factors. The factor loadings give the correlation between the newly found factor and the respective element. The higher the influence of an element for the factor, the higher is the factor loading. For all cores the following steps were performed before applying the PCA: (a) Using all measured values at 1 mm resolution results in a total sample amount of several thousands. Hence, data are resampled after smoothing using a Gaussian kernel filter with a 20 point window size to give a sample amount less than 1000. (b) Most of the parameters exhibited a pronounced skewness. PCA as well as Pearson correlation give unbiased results only for Gaussian distribution, which would be especially crucial for small data sets. However, the data were log transformed and znormalized to zero mean and unit standard derivation for each parameter separately in order to weight equally the different parameters in the multivariate analysis. (c) Subsequently, outliers were removed by deleting values greater/ less than mean plus/minus three time standard derivation of each measured element (includes 99.7% of all normal distributed values). The amount of significant and nontrivial new variables is measured using the scree test. Sulfur is not considered in the PCA since it correlates less well to the reference values and did not change the major findings of the PCA. However, S is known to be precipitated as pyrite and is thus a proxy for anoxic conditions of the bottom water. The concentration of an element in the lake sediment depends on the amount of influx and the degree of preservation of this element (ENGSTROM & WRIGHT 1984). Thus, the new calculated variables (principal component factors) are controlled by the same mechanisms of influx and preservation. For all numerical analysis MATLAB (TheMathworks, version 2009b) were used.

3.1 Calibration of µXRF results

The results of the µXRF scanning are presented as intensities (counts per second, cps). Since core DE3 is one of the most important cores of ELSA, discrete (and destructive) sample analysis using the WD-XRF is obtained to evaluate the intensity values given by the ED-XRF scanner with certain element concentrations (Tab. 1). For most elements (Al, Ti, Fe, Mn, Mg, Ca, Zr) good correlation (R2 > 0.75) can be observed between the intensities counted by the Eagle III µXRF and the corresponding measured concentrations (Tab. 2). The correlation was also checked for the standard-free quantification (fundamental parameter method) for weight percentages and atomic percentages. While the main elements fit quite well, Zr is the only trace element which shows an acceptable correlation between the measurements by WD-XRF and the intensities but does not have a significant correlation with the quantified results. However, Si matches better with the calculated weight percentages than with the intensity. The elements Cr, Ni, Cu, and Zn which show only weak correlation with the WD-XRF results are not further considered in this study. In the following the intensities are represented for the core SM3 and for DE3 weight% are shown because of good correlation between the ED- and WD-XRF results.

3.2 The elemental stratification

The comparison of the geochemical stratification (Fig. 4, 5) with the lithology of each single core leads to the suggestion, that Ca, Si and, K are the most prominent elements for the classification of aeolian sediments.

Core DE3 covers a time period from MIS3-MIS2, including the glacial inception of the late Weichselian ice sheets. However, single but severe dust events can be found between 43-41 m depth, which corresponds to the marine H4 event (HEMMING 2004), and beginning at 36 m the content of aeolian sediment is increasing, culminating in the last glacial maximum (LGM). Best correspondence with the lithological stratification can be found with the Si, Ca, and Zr. The stratification of the elements Si, Al, and Ti differ in the core DE3 (Fig. 4a, b). Si shows an overall increasing trend over the whole core section (from bottom to top), starting from around 60 wt.-% up to 70 wt.-%. The core section from 5–25 m has the highest content of Si with a slight increase between 15-25 m, while content in the core sections from 36-38 m and 58-60 m are decreases in relation to the overall trend and corresponding to an increase of organic matter in the sediment. In the core DE3 Ca is obviously enriched (CaO content around 5 wt.-%) in the upper section, with a depression between 10 and 20 m depth. Lower down core the content is around 1 wt.-% and reaches only higher values at 39 m, between 41-45 m, and in places between 59-68 m. The strong shift at 33 m depth from low to high amounts of Ca and Zr coincides with the last glacial ice advance of the late Weichselian, when the climate became colder and dryer and thus deflation processes became more frequent. Al and Ti are anti-correlated relating to Si and Mg and K are anti-correlated to Ca, showing a strong decrease between 33 and 25 m. Thus, these elements do not reflect a signal of allogenic aeolian input. Higher amounts of Al (around 15 wt.-%) correspond with the increase of gyttja and thus with an increase of chemical weathering due to more precipitation.

In the Holocene core SM3, the elements Si, Ti, Al, K and Ca are the most prominent ones to characterize loess gyttja and dust bearing sediment. Si, Al and Ti show corresponding stratigraphic patterns over the whole core (Fig. 5). Highest intensities of these elements are significant for the uppermost core meter and the core section 6.3-12.0 m. The latter is characterized by quite stable intensities of these geogenic elements, corresponding with loess gyttja. SM3 is characterized by a highly correlated stratification of Mg and K, corresponding to the signal of Ti and Al. Ca shows significant peaks. In core SM3 certain amounts of Ca are in the layer 5-5.5 m and 6.3-7 m depth. The sections 1-2 m and 7.5-10 m are also enriched in Ca. Al, Ti and Mg reaches the highest level in the uppermost meters. Si, K and Ca are enriched in the deepest part of the core and coincide with the highest dust occurrence of the core. In the core SM3 Zr shows a correlating trend to Si, Al and Ti, with maximum values around 2 m depth and from 6.5 m on down core. Sr peaks correspond to the major Ca peaks at 5 m and at 6.3 m. The latter peak represents the volcanic ash of the LST and shows strongly elevated values of the major geogenic elements, too.

In both cores the elements Fe, Mn, S and P match to core sections with higher amounts of organic matter and reflect environmental conditions with changing oxygenation of the bottom water or changes of the redox potential at the lake bottom. Over wide parts Fe and P correspond to the Al signal. S is mostly anticorrelated to these elements. This accords with vivianite concretions which are found in the core DE3 between 40 and 70 m depth. The mineral vivianite, $(Fe)_3(PO_4)_2$ *8H₂O, might be precipitated in absence of sulphur which would produce pyrite. In DE3 Fe, Mn and P have decreasing trends from the bottom to the top. Fe and Mn have corresponding trends, too. The Fe₂O₃(total) content increases from 10-5 wt.-% with an extraordinarily high value of nearly 20 wt.-% at 60 m depth. The MnO content is low and reaches a maximum of 1 wt.-% in 60 m, too. The highest values of S can be found at the same core sequence from 59 to 64 m depth. In this section the lost of ignition (LOI) is doubled from around 8 wt.-% to nearly 18 wt.-% (Tab. 1a). This section corresponds to the Greenland interstadial GIS-14, which is suggested to be the warmest interstadial of the last glacial cycle. The lowest levels of Fe and Mn are between 32 m depth until the top. A dip in their content is found between 40-46 m. Simultaneously Ca, Zr and S are enriched. In SM3 the elements Fe, Mn, P and S stratify the core from bottom to top into four parts and especially S is anti-correlated to the geogenic elements: (1) the deepest section (up to 7 m depth) is quite stable and shows only low counts of these elements. (2) Strong increase of S with decreasing Fe values and stable counts of Mn and P follows from 6.5-5.2 m. (3) In the sequence from 5.2 to 2.2 m all these elements correlate with each other and S shows its maximum values. Indeed, this is the section with the highest amount of organics in the core SM3. (4) In the uppermost section Fe, Mn and P correlate, and S is nearly anticorrelated to the other elements.

Tab. 1: WD-XRF results from the core DE3 from the Dehner dry maar, including main elements (upper panel) and trace elements (lower panel).Tab. 1: WD-XRF-Ergebnisse des Kerns DE3 aus dem Dehner Trockenmaar inklusive Hauptelemente (obere Tafel) und Spurenelemente (untere Tafel).

Tiefe [m]	LOI	SiO	2 ເນ [TiO2 A	l203	Fe2O3(t) (wt%)	Mn0 (wt%)	MgO (wt%)	CaO (wt%)	Na20	K20	P2	05 ·%1	S03 (wt%)	Sum ເ%ງ
4 50	7.05	742	- <u>-</u>		n 59	4 41		1 54	4.26	0.68	2 39	j (12	n 4n4	99.7
6.49	7.68	73.9	3	0.96 1	0.87	4.45	0.07	1.56	4.96	0.66	2.43	0.	11	0.472	100.5
8.70	7.45	73.5	8	1.02 1	1.39	4.81	0.07	1.64	4.52	0.67	2.49	0.3	13	0.452	100.8
10.60	6.14	78.7	0	1.08	8.05	3.40	0.06	1.33	4.72	0.74	1.98	0.:	11	0.422	100.6
14.50	6.56	75.8	7	0.98 1	.0.60	4.32	0.07	1.54	3.67	0.73	2.37	0.1	12	0.322	100.6
20.10	6.68	75.6	8	1.05 1	.0.80	4.21	0.06	1.49	3.58	0.73	2.31	0.1	11	0.225	100.3
21.78	6.33	79.1	.7	0.91	8.05	2.93	0.06	1.33	4.73	0.85	1.93	0.0	09	0.225	100.3
23.45	6.73	77.6	0	0.96	8.85	3.39	0.06	1.48	4.97	0.8	2.03	0.1	11	0.207	100.5
29.50	7.84	71.7	'8	0.99 1	.2.55	4.71	0.09	2.01	4.70	0.78	2.64	0.:	16	0.175	100.6
33.50	6.30	69.0	5	1.15 1	.6.16	6.23	0.11	2.11	1.51	0.72	3.19	0.3	18	0.062	100.5
39.55	6.41	67.8	4	1.25 1	.7.00	6.8	0.09	2.00	1.25	0.68	3.27	0.8	20	0.022	100.4
41.31	9.61	55.7	5	1.01 2	6.20	8.53	0.09	2.2	1.04	0.36	4.96	0.3	20	0.035	100.4
41.45	6.73	69.1	.4	1.16 1	.5.18	5.99	0.11	2.05	2.41	0.75	3.00	0.3	19	0.117	100.1
43.81	7.07	69.1	.6	1.17 1	.5.33	6.14	0.09	2.09	2.55	0.76	2.98	0.3	20	0.119	100.6
49.50	7.31	66.3	1	1.39 1	.7.90	7.44	0.09	2.09	1.16	0.64	3.20	0.3	27	0.022	100.5
52.50	7.20	66.0	0	1.55 1	.6.31	8.08	0.15	2.73	1.93	0.51	2.70	0.3	30	0.023	100.3
57.51	8.09	63.8	6	1.77 1	.8.05	9.30	0.16	2.16	1.39	0.46	2.86	0.3	35	0.011	100.4
59.57	16.98	59.1	.5	1.22 1	2.72	19.42	0.93	1.52	1.52	0.23	1.96	2.0	34	0.025	100.7
63.26	16.50	66.0	5	1.47 1	.4.57	10.33	0.45	1.91	1.69	0.27	2.26	1.4	46	0.086	100.6
68.47	9.05	60.6	0	1.48 2	0.83	9.11	0.18	2.53	1.08	0.47	3.70	0.3	30	0.03	100.3
68.63	10.05	56.7	1	1.37 1	.9.81	12.35	0.21	2.42	0.97	0.43	3.58	2.3	38	0.018	100.3
71.70	8.69	61.1	.9	1.44 2	0.59	8.66	0.11	2.43	1.04	0.48	3.74	0.3	36	0.03	100.1
77.88	6.75	62.6	1	1.35 1	.9.59	8.12	0.08	2.82	1.40	0.47	3.70	0.3	22	0.023	100.4
80.70	7.04	61.0	2	1.19 1	.9.84	8.12	0.15	2.54	2.03	0.48	3.79	0.9	92	0.01	100.1
83.31	10.19	61.6	4	1.44 2	0.10	8.07	0.08	2.84	1.83	0.45	3.57	0.3	26	0.043	100.3
Tiefe	V	Cr	Co	Ni	Cu	Zn	Rb	Sr	Y	Zr	Nb	Ba	Pb	La	Ce
Tiefe [m]	V (ppm)	Cr (ppm)	Co (ppm	Ni) (ppm)	Cu (ppm)	Zn) (ppm)	Rb (ppm)	Sr (ppm)	Y (ppm)	Zr (ppm)	Nb (ppm)	Ba (ppm)	Pb (ppr	n) (ppr	Ce n) (ppm)
Tiefe [m] 4.50	V (ppm) 79	Cr (ppm) 97	Co (ppm 13	Ni (ppm) 33	Cu (ppm) 18	Zn (ppm) 54	Rb (ppm) 90	Sr (ppm) 132	Y (ppm) 36	Z r (ppm) 545	Nb (ppm) 19	Ba (ppm) 379	Pb (ppm 21	n) (ppr 42	Ce n) (ppm) 85
Tiefe [m] 4.50 6.49	V (ppm) 79 80	Cr (ppm) 97 98	Co (ppm) 13 13	Ni (ppm) 33 31	Cu (ppm) 18 22	Zn (ppm) 54 62	Rb (ppm) 90 94	Sr (ppm) 132 142	Y (ppm) 36 37	Zr (ppm) 545 541	Nb (ppm) 19 20	Ba (ppm) 379 391	Pb (ppn 21 21	n) La (ppr 42 46	n) (ppm) 85 79
Tiefe [m] 4.50 6.49 8.70	V (ppm) 79 80 80	Cr (ppm) 97 98 103	Co (ppm) 13 13 15	Ni (ppm) 33 31 37 37	Cu (ppm) 18 22 34	Zn (ppm) 54 62 61	Rb (ppm) 90 94 96	Sr (ppm) 132 142 137	Y (ppm) 36 37 39	Zr (ppm) 545 541 551	Nb (ppm) 19 20 19	Ba (ppm) 379 391 401	Pb (ppn 21 21 19	h La n) (ppr 42 46 41	n) Ce (ppm) 85 79 84
Tiefe [m] 4.50 6.49 8.70 10.60	V (ppm) 79 80 80 60	Cr (ppm) 97 98 103 110	Co (ppm) 13 13 15 8	Ni (ppm) 33 31 37 24	Cu (ppm) 18 22 34 13	2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	Rb (ppm) 90 94 96 69	Sr (ppm) 132 142 137 134	Y (ppm) 36 37 39 45	Zr (ppm) 545 541 551 956	Nb (ppm) 19 20 19 22 22	Ba (ppm) 379 391 401 344	Pb (ppn) 21 21 19 20 18	La n) (ppr 42 46 41 48 48 447	Ce (ppm) 85 79 84 84
Tiefe [m] 4.50 6.49 8.70 10.60 14.50	V (ppm) 79 80 80 60 78 78	Cr (ppm) 97 98 103 110 105	Co (ppm) 13 13 15 8 13	Ni (ppm) 33 31 37 24 34	Cu (ppm) 18 22 34 13 15	Zn (ppm) 54 62 61 46 57	Rb (ppm) 90 94 96 69 91	Sr (ppm) 132 142 137 134 128	Y (ppm) 36 37 39 45 38 (11)	Zr (ppm) 545 541 551 956 615	Nb (ppm) 19 20 19 20 20 20 20 20 20 20 20 20 20 20 20 20 20 20	Ba (ppm) 379 391 401 344 367	Pb (ppn 21 21 21 21 19 20 18 10	La n) (ppr 42 46 44 46 44 47 47 47	Ce (ppm) 85 79 84 90 92
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10	V (ppm) 79 80 80 60 78 77 77	Cr (ppm) 97 98 103 110 105 104	Co (ppm) 13 13 15 8 13 11	Ni (ppm) 33 31 37 24 34 30	Cu (ppm) 18 22 34 13 15 16	Zn (ppm) 54 62 61 46 57 58 41	Rb (ppm) 90 94 96 69 91 90 70	Sr (ppm) 132 142 137 134 128 131	Y (ppm) 36 37 39 45 38 41 28	Zr (ppm) 545 541 551 956 615 649 721	Nb (ppm) 19 20 19 20 20 20 20 20 20 19 20 20 20 20 20 20 20 20 20	Ba (ppm) 379 391 401 344 367 387	Pb (ppn 21 21 19 20 18 19 10	La (ppr 42 46 41 48 47 50	Ce [ppm] 85 79 84 90 87
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 22.45	V (ppm) 79 80 80 60 78 77 48	Cr (ppm) 97 98 103 103 105 104 91	Co (ppm) 13 13 15 8 13 11 4	Ni (ppm) 33 31 37 24 34 30 22 34 30 20	Cu (ppm) 18 22 34 13 15 16 10	Zn (ppm) 54 62 61 46 57 58 41	Rb (ppm) 90 94 96 91 90 91 90 70	Sr (ppm) 132 142 137 134 128 131 141	Y (ppm) 36 37 39 45 38 41 38 20	Zr (ppm) 545 541 551 956 615 649 721	Nb (ppm) 19 20 19 20 20 20 19 19 19 19 20 19 21 22 20 19 20 20 19 19	Ba (ppm) 379 391 401 344 367 387 347	Pb (ppn) 21 21 19 20 18 19 16	La n) (ppr 42 46 41 48 47 50 51 51	Ce [ppm] 85 79 84 90 87 90 87 90 87 90 87 90 87
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50	V (ppm) 79 80 60 78 77 48 774 80 77 80 77 80	Cr (ppm) 97 98 103 110 105 104 91 99	Co (ppm 13 13 15 8 13 11 4 4 8	Ni (ppm) 33 31 37 24 34 30 20 20 29	Cu (ppm) 18 22 34 13 15 16 10 10 14	Zn (ppm) (ppm) 54 62 61 46 257 58 58 41 41 48	Rb (ppm) 90 94 96 91 90 91 90 91 90 91 90 91 90 91 90 91 90	Sr (ppm) 132 142 137 134 128 131 141 141 143	Y (ppm) 36 37 39 45 38 41 38 39 37	Zr (ppm) 545 541 551 956 615 649 721 681 (169	Nb (ppm) 19 20 19 20 19 20 19 20 19 20 19 20 19 20 19 20 19 19 20 20 20 19 20	Ba (ppm) 379 391 401 344 367 387 347 337 402	Pb (ppn 21 21 19 20 18 19 16 16	La (ppr 42 46 41 48 47 50 51 42	Ce (ppm) 85 79 84 84 90 87 79 82 82
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 22.50	V (ppm) 380 80 60 78 77 48 59 59 88	Cr (ppm) 97 98 103 110 105 104 91 99 105	Co (ppm) 13 15 8 13 13 11 4 8 11 4 8 14	Ni (ppm) 33 31 37 24 34 20 29 40 53	Cu (ppm) 18 22 34 13 15 16 10 14 23 24	Zn (ppm) 54 62 61 46 57 58 41 48 64	Rb (ppm) 90 94 96 91 90 91 90 70 70 105 125	Sr (ppm) 132 142 137 134 128 131 141 143 154 154	Y (ppm) 36 37 39 45 38 41 38 39 37	Zr (ppm) 545 541 551 956 615 649 721 681 469 368	Nb (ppm) 19 20 19 20 19 20 19 20 19 20 19 20	Ba (ppm) 379 391 391 344 367 387 387 387 344 525	Pb (ppn 21 21 19 20 18 19 16 21 21	La n) (ppr 42 46 41 48 47 50 51 42 42 43	Ce (ppm) 85 79 84 90 87 79 84 90 87 88 87 88 88 88 88 88 88 88 88 88 88 88
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55	V (ppm) 30 80 60 78 77 48 59 88 122 137	Cr (ppm) 97 98 103 105 104 91 99 105 120 120	Co (ppm) 13 15 8 13 11 4 4 8 14 20 24	Ni (ppm) 33 31 37 24 34 20 29 40 53 60	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39	Zn (ppm) 54 62 61 46 57 58 41 48 64 81	Rb (ppm) 90 94 96 97 90 91 90 91 90 105 135 140	Sr (ppm) 132 142 137 134 128 131 141 143 144 143 154 160 142	Y (ppm) 36 37 39 45 38 41 38 39 37 38 39 37 38 39 37 38 37	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341	Nb (ppm) 19 20 19 20 19 20 19 20 19 20 19 20 20 20 20 20 20 19 20	Ba (ppm) 379 391 401 344 367 387 347 337 347 535 505	Pb (ppn 21 21 19 20 18 19 16 16 21 23 25	La n) (ppr 42 46 41 48 47 50 51 42 43 60 46 46	Ce (ppm) 85 79 84 90 87 90 87 99 82 99 107
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31	 V (ppm) 79 80 60 78 77 48 59 88 122 137 195 	Cr (ppm) 97 98 103 110 105 91 99 105 120 121 121	Co (ppm) 13 15 8 13 13 11 4 8 14 20 24 17	Ni (ppm) 33 31 37 24 34 220 29 40 53 60 71	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39 58	Zn (ppm) 54 62 61 46 57 58 41 64 64 81 85 99	Rb (ppm) 90 94 96 91 90 91 90 105 135 140 224	Sr (ppm) 132 142 137 134 128 131 141 143 154 160 142 153	Y (ppm) 36 37 39 45 38 41 38 39 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 29	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183	Nb (ppm) 19 20 19 20 19 20 19 20 23 28 31	Ba (ppm) 379 391 401 344 367 387 387 347 357 505 505	Pb (ppn) 21 19 200 18 19 16 16 21 23 25 33	La n) (ppr 42 46 41 48 47 50 51 42 43 60 466 59	Ce [ppm] 85 79 84 90 87 90 87 99 107 119
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45	 V (ppm) 79 80 60 78 77 48 59 88 122 137 195 117 	Cr (ppm) 97 98 103 110 105 91 99 105 120 121 121 149	Co (ppm) 13 15 8 13 11 4 8 11 4 8 14 20 24 17 21	Ni (ppm) 33 31 37 24 34 30 24 34 30 20 29 40 53 60 71 52	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39 58 40	Zn (ppm) 54 62 61 46 57 46 47 48 64 81 85 99 76	Rb (ppm) 90 94 96 91 90 91 90 70 70 105 135 140 224 126	Sr (ppm) 132 142 137 134 128 131 141 143 154 153 150	Y (ppm) 36 37 39 45 38 41 38 39 37 38 39 37 38 39 37 38 37 38 37 38 37 29 37	Zr (ppm) 545 547 551 956 615 649 721 681 368 341 183 380	Nb (ppm) 19 20 19 20 19 20 19 22 20 19 22 20 20 20 20 20 20 20 20 20 20 20 23 28 31 26	Ba (ppm) 379 391 391 344 367 347 367 387 397 397 397 397 397 397 397 397	Pb (ppm 21 19 200 18 19 20 18 19 20 18 19 20 18 19 20 18 19 16 21 23 25 33 21	La n) (ppr 42 46 41 48 47 50 51 42 43 60 46 59 49 49	Ce (ppm) 85 79 84 90 87 90 87 90 107 119 102
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81	V (ppm) 30 30 60 78 77 48 59 48 59 88 122 137 137 195 117	Cr (ppm) 97 98 103 105 104 91 99 105 120 121 121 149 113	Co (ppm) 13 15 8 13 11 4 8 14 20 24 17 21	 Ni (ppm) 33 31 37 24 34 26 20 20 20 20 53 60 53 60 71 52 54 	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39 58 40 39	Zn (ppm) 54 62 61 46 57 46 41 48 64 81 85 99 76 78	Rb (ppm) 90 94 95 96 91 90 91 90 91 92 91 90 91 92 93 94 95 91 92 105 135 140 224 126 125	Sr [ppm] 132 142 137 134 128 131 141 143 154 160 142 153 150 164	Y (ppm) 36 37 39 45 38 41 38 39 37 38 39 37 38 37 38 37 38 37 38 37 38 37 29 37 38	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379	Nb (ppm) 19 20 19 20 19 20 19 20 20 20 20 20 20 20 20 20 20 20 31 26 28 21	Ba (ppm) 379 391 391 344 367 387 387 387 387 387 387 387 387 387 387 387 407 535 605 497 486 532	Pb (ppm 21 19 20 18 19 20 18 19 20 18 19 20 18 19 20 18 19 20 18 19 20 18 19 16 16 21 23 25 33 21 21	La n) (ppr 42 46 41 48 47 50 51 42 43 60 46 59 49 50	Ce [ppm] 85 79 84 90 87 90 87 90 107 119 102 104
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81	 V (ppm) 80 80 77 78 77 48 59 88 122 137 195 117 121 150 	Cr (ppm) 97 98 103 105 104 99 99 105 120 120 121 121 149 113	Co (ppm) 13 15 8 13 11 4 8 14 20 24 17 21 17 21	Ni (ppm) 33 31 37 24 34 20 29 40 53 60 71 52 54 62	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 23 34 39 58 40 39 54	Zn (ppm) 54 62 61 46 57 58 41 64 64 81 99 76 78 91	Rb (ppm) 90 94 96 97 90 91 90 91 92 105 135 140 125 125 124	Sr (ppm) 132 142 137 134 128 131 141 143 143 154 160 142 153 150 164 184	Y (ppm) 36 37 39 45 38 41 38 39 37 38 39 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 36	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379 313	Nb (ppm) 19 20 19 20 19 20 21 20 20 20 20 20 20 20 20 20 20 20 20 20 20 20 20 31 26 28 37	Ba (ppm) 379 391 401 344 367 347 347 347 357 347 357 347 357 407 535 505 497 532 532 623	Pb (ppm 211 19 200 188 199 200 188 199 166 211 233 253 333 211 221	La n) (ppr 42 46 411 48 47 50 51 42 43 60 59 49 50 50 49 50	Ce (ppm) 85 79 84 90 87 90 87 90 87 107 119 102 104
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81 49.50 52.50	 V (ppm) 79 80 60 78 77 48 59 88 122 137 195 117 121 160 179 	Cr (ppm) 97 98 103 110 105 99 105 120 121 121 121 121 121 121 121 121 121	Co (ppm) 13 15 8 13 11 4 8 14 20 24 17 21 17 21 17 23 26	Ni (ppm) 33 31 37 24 34 220 29 40 53 60 71 52 54 52 54 62 87	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39 58 40 39 54 105	Zn (ppm) 54 62 61 46 57 58 41 64 64 81 85 99 76 78 91 91	Rb (ppm) 90 94 96 91 90 91 90 91 92 93 105 135 140 224 125 149 126 127	Sr (ppm) 132 142 137 134 128 131 141 143 154 160 142 153 150 164 184 189	Y (ppm) 36 37 39 45 38 41 38 31 38 39 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 36 35	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379 313	Nb (ppm) 19 20 19 20 19 20 20 20 20 20 20 20 20 20 20 20 20 31 26 28 31 26 28 31 26 28 37 42	Ba (ppm) 379 391 401 344 367 387 387 347 387 387 387 347 367 387 387 407 535 497 486 532 623 623 623	Pb (ppn 211 19 200 188 199 166 21 233 255 333 21 21 23 255 333 21 24 24	La Im (ppr 42 46 41 48 47 50 51 42 43 60 466 59 49 50 50 50 49 50 62 57	Ce (ppm) 85 79 84 90 87 90 87 90 87 107 119 102 104 115 117
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81 49.50 52.50 57.51	 V (ppm) 79 80 60 78 77 48 59 88 122 137 195 117 121 160 179 201 	Cr (ppm) 97 98 103 110 105 99 99 105 120 121 120 121 120 121 120 121 120 121 120 121 120	Co (ppm) 13 15 8 13 11 4 4 8 14 20 24 17 21 17 21 17 23 26 26 26	Ni (ppm) 33 31 37 24 34 22 34 30 24 34 32 40 29 40 53 60 71 52 54 62 87 80	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39 58 40 39 58 40 39 554 105 71	Zn (ppm) 54 62 61 46 57 58 41 64 64 81 85 99 76 78 91 92	Rb (ppm) 90 94 96 91 90 91 90 91 90 105 135 140 224 125 149 125 143	Sr (ppm) 132 142 137 134 128 131 141 143 154 160 142 153 150 164 184 189 180	Y (ppm) 36 37 39 45 38 41 38 317 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 36 35 38	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379 313 303 345	Nb (ppm) 19 20 19 20 19 20 21 22 31 26 37 42 51	Ba (ppm) 379 391 391 344 367 347 367 387 391 391 391 391 392 497 486 532 623 667 705	Pb (ppm 21 19 200 18 19 200 18 19 201 18 19 200 18 19 16 16 21 23 25 33 21 21 21 21 21 21 21 21 21 21 21 22	La n) (ppr 42 46 41 48 47 50 51 42 43 60 46 59 49 50 60 59 49 50 62 57 63 57	Ce (ppm) 85 79 84 90 87 90 87 90 87 90 87 107 102 102 104 115 117 118
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81 49.50 52.50 57.51 59.57	V (ppm) 30 30 30 77 78 77 48 59 88 122 137 121 137 121 121 121 121 120 120 179 201	Cr (ppm) 97 98 103 105 104 99 105 120 120 121 121 121 121 123 123 123 123 123 123	Co (ppm) 13 15 8 13 11 4 8 14 20 24 17 21 17 21 17 23 26 26 26 22	Ni (ppm) 33 31 37 24 37 24 37 24 37 24 37 24 37 24 34 37 24 34 37 24 34 37 29 40 53 60 71 52 54 62 87 80 76	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39 58 40 39 54 105 71 74	Zn (ppm) 54 62 61 46 57 46 61 46 85 99 76 91 92 92 91 92 91 92 91 92 91	Rb (ppm) 90 94 96 91 90 91 90 91 92 105 135 140 224 126 127 136 127 136 99	Sr [ppm] 132 142 137 134 128 131 142 131 142 134 154 160 142 153 150 164 184 189 134	Y (ppm) 36 37 39 45 38 41 38 39 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 36 35 38 24	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379 313 303 345 166	Nb (ppm) 19 20 19 20 19 20	Ba [ppm] 379 391 391 344 367 387	Pb (ppm 21 19 20 18 19 20 18 19 20 18 19 20 18 19 20 18 19 16 16 21 23 25 33 21 24 24 25 22	La n) (ppr 42 46 41 48 47 50 51 42 43 60 43 60 59 49 50 50 62 57 63 49	Ce (ppm) 85 79 84 90 87 90 87 99 107 119 102 104 115 117 118 99
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81 49.50 52.50 57.51 59.57 63.26	 V (ppm) 79 80 60 78 77 48 59 88 122 137 155 117 121 160 179 201 220 260 	Cr (ppm) 97 98 103 105 104 99 105 120 120 121 121 129 123 120 121 129 123 120 120 120 120 120 120 120 120 120 120	Co (ppm) 13 15 8 13 11 4 8 14 20 24 17 21 17 21 17 23 26 26 26 22 30	Ni (ppm) 33 31 37 24 37 24 37 24 34 37 24 34 37 24 34 37 24 34 30 20 29 40 53 60 71 52 54 62 87 80 76 82	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39 58 40 39 54 105 71 74 93	Zn (ppm) 54 62 61 46 57 58 41 64 64 81 99 76 99 76 91 92 91 92 91 107	Rb (ppm) 90 94 96 91 90 91 90 91 92 105 1264 1255 140 1264 127 1386 149 127 1386 99 127 1386 99 1155	Sr (ppm) 132 142 137 134 128 131 142 131 142 134 143 144 143 154 160 142 153 150 164 184 189 134 135 158	Y (ppm) 36 37 39 45 38 41 38 39 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 36 35 38 24 26	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379 313 303 345 166 187	Nb (ppm) 19 20 19 20 19 20	Ba (ppm) 379 391 391 344 367 347 347 347 357 347 357 347 357 347 357 3407 535 505 407 535 505 497 486 532 623 667 705 568 735	Pb (ppm 211 19 200 188 199 200 188 199 166 211 233 253 333 211 244 255 222 233	La Im (ppr 42 46 411 48 47 50 50 51 42 43 60 46 59 49 50 50 49 50 62 57 63 49 66 68	Ce (ppm) 85 79 84 90 87 90 87 90 107 119 102 104 115 117 118 99 114
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81 49.50 52.50 57.51 59.57 63.26 68.47	 V (Ppm) 79 80 60 78 77 48 59 48 122 137 195 117 121 160 179 201 220 260 215 	Cr (ppm) 97 98 103 105 104 99 105 120 121 120 121 121 120 121 121 121 125 120 121 125 120 121 120 121 120 121 120 121 120 120	Co (ppm) 13 15 8 13 11 4 8 14 20 24 17 21 17 21 17 23 26 26 26 22 30 36	Ni (ppm) 33 31 37 24 37 24 37 24 37 24 34 37 24 34 37 24 34 37 24 34 37 40 20 40 53 60 71 52 54 62 87 80 76 82 88	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39 58 40 39 54 105 71 74 93 87	Zn (ppm) 54 62 61 46 57 58 41 48 64 81 85 99 76 91 92 91 92 91 107 120	Rb (ppm) 90 94 95 94 95 97 90 70 70 73 105 135 140 224 125 1420 125 140 126 127 136 99 115 170	Sr (ppm) 132 142 137 134 128 131 141 143 154 160 142 153 150 164 189 180 189 180 134 158	Y (ppm) 36 37 39 45 38 41 38 31 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 36 35 38 24 26 34	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379 313 303 345 166 187 222	Nb [ppm] 19 20 19 20 19 20	Ba (ppm) 379 391 401 367 367 387 387 344 367 387	Pb (ppn 211 19 200 188 199 166 21 233 255 333 21 231 255 333 21 244 254 223 355	La Im (ppr 42 46 41 48 47 50 51 42 43 60 46 60 59 49 50 50 49 50 62 57 63 49 66 65	Ce (ppm) 85 79 84 90 84 90 87 90 107 119 102 104 115 117 118 99 114
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81 49.50 52.50 57.51 59.57 63.26 68.47 68.63	V (ppm) 79 80 60 77 48 77 48 177 48 122 137 195 117 121 160 179 201 220 260 215 199	Cr (ppm) 97 98 103 110 105 99 99 105 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 120	Co (ppm) 13 15 8 13 11 4 8 14 20 24 17 21 17 23 26 26 26 26 22 30 36 33	Ni (ppm) 33 31 37 24 37 24 37 24 37 24 34 37 24 34 37 24 34 37 24 34 37 40 29 40 53 60 71 52 54 62 54 62 87 80 76 82 88 79	Cu (ppm) 18 22 34 13 15 16 100 14 23 34 39 58 400 39 58 40 39 58 40 39 58 40 39 58 40 39 58 40 39 58 40 39 58 40 39 54 105 71 74 93 87 67	Zn (ppm) 54 62 61 46 57 58 41 64 81 85 99 76 91 91 92 91 107 120 112	Rb (ppm) 90 94 96 91 90 91 90 91 90 91 92 93 105 135 140 224 126 125 149 127 136 99 115 170 158	Sr (ppm) 132 142 137 134 128 131 142 137 142 137 142 143 144 143 154 160 142 153 150 164 189 180 134 158 151 148	Y (ppm) 36 37 39 45 38 41 38 317 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 36 37 38 36 37 38 36 38 38 38 38 38 38 38 38 39 38 39 38	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379 313 303 345 166 187 222 205	Nb [ppm] 19 20 19 20 19 20 21 22 31 26 31 26 37 42 53 40 39	Ba (ppm) 379 391 401 367 367 387 387 344 367 387	Pb (ppn 211 19 200 18 199 16 16 21 23 25 333 211 24 24 25 23 35 30	La Im [ppr 42 46 411 48 47 50 51 42 43 60 466 59 49 50 62 57 63 49 66 65 55 56	Ce (ppm) 85 79 84 90 87 90 87 90 87 90 107 102 102 104 115 117 118 99 114 140 118
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81 49.50 52.50 57.51 59.57 63.26 68.47 68.63 71.70	 V (ppm) 79 80 60 78 77 48 59 88 122 137 195 117 121 160 179 201 220 260 215 199 205 	Cr (ppm) 97 98 103 110 105 99 99 105 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 120	Co (ppm) 13 15 8 13 11 4 8 14 20 24 17 21 17 21 17 23 26 26 26 26 22 30 36 33 32	Ni (ppm) 33 31 37 24 37 24 37 24 37 24 37 24 37 24 34 37 24 34 37 24 37 20 20 20 30 30 30 31 32 33 34 35 36 37 38 39 39 39 39 39 39 39 39 39 39 39 39 39 39 39	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39 58 40 39 54 105 71 74 93 87 67 75	Zn (ppm) 54 62 61 46 57 46 67 85 99 76 91 91 92 91 107 120 112	Rb (ppm) 90 94 96 91 90 70 70 105 126 1224 126 127 136 127 136 127 136 99 127 136 99 115 158 168	Sr [ppm] 132 142 137 134 128 131 142 137 134 154 160 142 153 150 164 184 189 134 158 151 148 189 134 158 134 158 134	Y (ppm) 36 37 39 45 38 41 38 31 38 39 37 38 37 38 37 38 37 38 37 38 37 38 36 35 38 24 26 34 34	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379 313 303 345 166 187 222 205 228	Nb [ppm] 19 20 19 20 19 20 20 20 20 20 20 20 20 20 20 20 20 31 26 31 26 37 42 51 42 53 40 39 37	Ba [ppm] 379 391 391 344 367 387	Pb (ppm 21 19 20 18 19 20 18 19 20 18 19 20 18 19 16 16 21 23 21 24 24 25 22 23 35 300 33	La n] (ppr 42 46 41 48 47 50 51 42 43 60 43 60 49 50 62 57 63 62 49 66 55 56	Ce (ppm) 85 79 84 90 87 90 87 90 87 90 107 119 102 104 115 117 118 99 114 140 118 131
Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81 49.50 52.50 57.51 59.57 63.26 68.43 71.70 77.88	 (ppm) 79 80 60 78 77 48 59 88 122 137 157 121 160 179 201 220 260 215 199 205 176 	Cr (ppm) 97 98 103 105 104 99 105 120 121 120 121 121 121 123 121 125 120 125 120 125 120 125 120 125 120 125 120 125	Co (ppm) 13 15 8 13 11 4 8 14 20 24 17 21 17 21 17 23 26 26 26 26 26 22 30 36 33 32 30	Ni (ppm) 33 31 37 24 37 24 37 24 34 27 40 29 40 53 40 53 40 53 40 53 40 53 40 53 40 53 40 53 40 53 60 71 52 54 62 87 80 76 88 79 82 79 82 79	Cu [ppm] 18 22 34 13 15 16 10 14 23 34 39 58 40 39 54 105 71 74 93 87 67 75 74	Zn (ppm) 54 62 61 46 57 46 67 85 64 85 99 76 91 91 92 91 107 120 112 113	Rb (ppm) 90 94 96 91 90 91 90 105 126 1224 126 127 138 140 126 127 138 149 127 138 149 127 138 149 127 138 149 127 138 149 127 138 149 128 149 128 138 139 1315 149 158	Sr (ppm) 132 142 137 134 128 131 142 131 142 134 143 144 143 154 160 142 153 150 164 189 180 134 158 151 148 158 151 248 233	Y (ppm) 36 37 39 45 38 41 38 317 38 391 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 37 38 36 35 38 24 26 34 34 35	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379 313 303 445 186 379 313 303 222 205 228 238	Nb (ppm) 19 20 19 20 19 20 21 22 23 37 32	Ba [ppm] 379 391 391 307 344 367 347 367 347 367 347 367 347 357 407 535 505 497 486 532 623 667 705 568 735 625 612 567 567 612 567 625 625 625 625 625	Pb (ppm 21 19 20 18 19 20 18 19 20 18 19 20 18 19 20 18 19 16 16 21 23 25 33 21 24 25 22 23 35 30 33 30	La I Ippr 42 42 411 48 47 50 42 43 43 60 43 60 59 49 50 50 49 50 49 50 66 57 63 66 56 56 70 69	Ce (ppm) 85 79 84 90 84 90 87 90 87 99 107 119 102 104 115 117 118 99 114 140 118 131 131
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Tiefe [m] 4.50 6.49 8.70 10.60 14.50 20.10 21.78 23.45 29.50 33.50 39.55 41.31 41.45 43.81 49.50 52.50 57.51 59.57 63.26 68.47 68.63 71.70 77.88 80.70 83.31	V 79 80 60 77 48 77 48 77 48 177 48 122 137 195 117 201 201 220 260 215 199 205 176 205 176 284	Cr (ppm) 97 98 103 110 105 99 105 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 121 120 120	Co (ppm) 13 15 8 13 11 4 8 11 4 8 14 20 24 17 21 17 21 17 23 26 26 26 26 26 22 30 36 33 32 30 32 30 27 29	Ni (ppm) 33 31 37 24 37 24 37 24 34 37 24 34 37 24 34 37 24 34 37 24 37 24 37 40 20 40 53 60 71 52 54 62 54 62 87 88 79 82 79 82 79 82 79 76 75	Cu (ppm) 18 22 34 13 15 16 10 14 23 34 39 58 40 39 58 40 39 58 40 39 58 40 39 58 40 39 58 40 39 58 40 39 58 71 74 62 102	Zn (ppm) 54 62 61 46 57 58 41 64 64 81 76 99 76 91 91 91 91 91 107 120 112 113 96 110	Rb (ppm) 90 94 96 91 90 91 90 91 90 91 90 91 90 91 105 126 1224 126 127 136 127 136 99 115 1700 158 168 1564	Sr (ppm) 132 142 137 134 128 131 142 137 142 137 142 143 144 143 154 160 142 153 150 164 189 180 134 155 151 148 151 148 153 255 174	Y (ppm) 36 37 39 45 38 41 38 317 38 37 38 37 38 37 38 37 38 37 38 37 38 36 35 38 24 26 34 34 34 35 34 35 34	Zr (ppm) 545 541 551 956 615 649 721 681 469 368 341 183 380 379 313 303 345 166 187 222 205 238 228 238 226 238 228 238 226 238 228 238	Nb 19 20 19 20 19 20 21 22 23 24 37 42 53 40 39 37 33 29 45	Ba (ppm) 379 391 401 367 387 387 344 367 387	Pb (ppn 21 19 200 18 199 16 16 21 233 255 333 21 24 24 24 24 24 24 24 24 24 24 24 25 22 335 300 333 30 333 30 333	La Im [ppr 42 44 44 44 47 50 42 43 43 40 43 40 43 50 46 59 49 50 62 57 63 49 66 65 56 70 69 56 76 56	Ce Ce (ppm) 85 79 84 90 87 90 87 90 87 90 107 119 102 117 118 99 114 118 99 114 123 123 123 123 123 123

Tab. 2: Correlation of DE3 µXRF results (Intensity, Wt%, At%) with WD-XRF analysis. In general, the main elements show the highest correlations, at least in comparison with the measured intensity values with the WD-XRF results. Italic printed values are not significant at the 0.05 level. Highest values of correlation are highlighted with bold letters.

Tab. 2: Korrelationsergebnisse (DE3) der µXRF-Resultate (Intensität, Gew.-%, Atom.-%) mit den WD-XRF-Analysen an diskreten Proben. Höchste Korrelationen sind hervorgehoben. Kursiv gedruckte Werte sind auf dem Niveau von 0,05 nicht signifikant. Höchste Korrelationen sind fett hervorgehoben.

	SiO2 (Wt%) WD	Si - Int.	Si - Wt%	Si - At%
SiO2 (%) WD	1.00			
Si - Int.	0.41	1.00		
Si - Wt%	0.69	0.81	1.00	
Si - At%	0.53	0.74	0.97	1.00

	TiO2 (Wt%) WD	Ti - Int.	Ti - Wt%	Ti - At%
TiO2 (%) WD	1.00			
Ti - Int.	0.84	1.00		
Ti - Wt%	0.70	0.86	1.00	
Ti - At%	0.71	0.82	0.98	1.00

	Na20 (Wt %) WD	Na - Int.	Na - Wt%	Na - At%
Na20 (%) WD	1.00			
Na - Int.	0.41	1.00		
Na - Wt%	0.20	0.11	1.00	
Na - At%	0.17	0.10	1.00	1.00

	Cr (ppm) WD	Cr - Int.	Cr - Wt%	Cr - At%
Cr (ppm) WD	1.00			
Cr - Int.	-0.32	1.00		
Cr - Wt%	-0.31	0.28	1.00	
Cr - At%	-0.32	0.33	0.98	1.00

	Al2O3 (Wt%) WD	Al – Int.	Al - Wt%	Al - At%	
Al203 (%) WD	1.00				
Al - Int.	0.66	1.00			ſ
Al - Wt%	0.74	0.95	1.00		ſ
Al - At%	0.82	0.91	0.98	1.00	

	Fe2O3(t) (Wt%) WD	Fe - Int.	Fe - Wt%	Fe - At%
Fe2O3(t) (%) WD	1.00			
Fe - Int.	0.95	1.00		
Fe - Wt%	0.95	0.85	1.00	
Fe - At%	0.93	0.82	1.00	1.00

	MnO (Wt%) WD	Mn - Int.	Mn - Wt%	Mn - At%
Mn0 (%) WD	1.00			
Mn - Int.	0.99	1.00		
Mn - Wt%	0.98	0.98	1.00	
Mn - At%	0.98	0.97	1.00	1.00

	MgO (Wt%) WD	Mg - Int.	Mg - Wt%	Mg - At%
Mg0 (%) WD	1.00			
Mg - Int.	0.82	1.00		
Mg - Wt%	0.72	0.77	1.00	
Mg - At%	0.79	0.81	0.99	1.00

	CaO (Wt%) WD	Ca - Int.	Ca - Wt%	Ca - At%
CaO (%) WD	1.00			
Ca - Int.	0.98	1.00		
Ca - Wt%	0.96	0.98	1.00	
Ca - At%	0.96	0.98	1.00	1.00

Ni (ppm) WD	Ni - Int.	Ni - Wt%	Ni - At%
1.00			
-0.21	1.00		
-0.03	0.38	1.00	
0.10	0.37	0.94	1.00
	Ni (ppm) WD 1.00 -0.21 -0.03 0.10	Ni (ppm) WD Ni - Int. 1.00 1.00 -0.21 1.00 -0.03 0.38 0.10 0.37	Ni (ppm) WD Ni - Int. Ni - Wt% 1.00 -0.03 -0.38 -0.03 0.38 1.00 0.10 0.37 0.94

	Cu (ppm) WD	Cu - Int.	Cu - Wt%	Cu - At%
Cu (ppm) WD	1.00			
Cu - Int.	-0.24	1.00		
Cu - Wt%	-0.18	0.05	1.00	
Cu - At%	-0.05	-0.15	0.95	1.00

	Zn (ppm) WD	Zn - Int.	Zn - Wt%	Zn - At%
Zn (ppm) WD	1.00			
Zn - Int.	-0.35	1.00		
Zn - Wt%	-0.10	0.60	1.00	
Zn - At%	-0.01	0.46	0.94	1.00

	Sr (ppm) WD	Sr - Int.	Sr - Wt%	Sr - At%
Sr (ppm) WD	1.00			
Sr - Int.	0.17	1.00		
Sr - Wt%	-0.34	0.28	1.00	
Sr - At%	-0.37	0.18	0.98	1.00

	Zr (ppm) WD	Zr - Int.	Zr - Wt%	Zr - At%
Zr (ppm) WD	1.00			
Zr - Int.	0.78	1.00		
Zr - Wt%	0.39	0.02	1.00	
Zr - At%	0.24	-0.10	0.96	1.00



Fig. 4: Continuous measured element concentrations of the core DE3 in (upper panel) cps and (lower panel) quantified (expressed as wt.-% and ppm) using fundamental parameter method (smoothed by a 50-point running mean filter) and measured with the Eagle III µXRF. Black crosses are corresponding results from WD-XRF, measured on discrete samples. Apart from Al all energy dispersive measured elements are underrepresented but show clearly covarying trends. The Ca values represent a dust signal (DIETRICH & SIROCKO 2009).

Abb. 4: Mit der μ RFA kontinuierlich gemessene Elementkonzentrationen des Sedimentkerns DE3, angegeben in (upper panel) counts per second (cps) und (lower panel) quantifiziert mittels der Fundamentalparametermethode in Gew.-% (ppm). Die Werte wurden durch einen gleitenden Mittelwert (Fensterbreite 50-Punkt) geglättet. Die schwarzen Kreuze entsprechen den Vergleichsmessungen an diskreten Proben, die mit einer WD-RFA bestimmt wurden. Mit Ausnahme von Aluminium, zeigen die energie-dispersiv gemessenen Werte niedrigere Konzentrationen an. Insgesamt zeigt die Elementverteilung mit beiden Methoden aber eine deutliche Kovarianz. Hohe Kalziumwerte weisen auf eine verstärkte Staubaktivität hin (DIETRICH & SIROCKO 2009).



Fig. 5: Continuous measured element concentrations of the core SM3 measured in cps (smoothed by a 5-point running mean filter). The distinctive split in a glacial (>7 m depth) and an interglacial part is clearly visible in the element concentrations of minerogenous elements. The LST at 6.3 m depth is characterized by certain peaks of Si, Ti, Al, and K. Abb. 5: Kontinuierlich gemessene Elementkonzentrationen des Sedimentkerns SM3 (in cps) und geglättet durch einen gleitenden Mittelwert (Fensterbreite 5-Punkt). Deutliche Unterschiede in der geochemischen Zusammensetzung sind vor allem beim Vergleich kaltzeitlicher Sedimente (> 7 m Tiefe) mit warmzeitli $cher \ und \ organikreicher \ Sedimente$ zu sehen. Erstere sind durch deutlich höhere Konzentration minerogener Elemente gekennzeichnet. Die LST in 6,3 m Tiefe zeigt deutliche Spitzenwerte von Si, Ti, Al und K.

Tab. 3: Upper panel: Correlation matrix of main elements (core DE3 measured intensities by μ XRF). Italic printed values are not significant at the 0.05 level. Lower panel: Factor loadings after PCA calculation of main elements (core DE3 measured intensities by μ XRF), including Eigenwert (Eigen.), the described variance (dVar.) and the cumulative sum of dVar (% Cumul). Bold printed values ($\geq |0.4|$) have a major influence on the single principle components.

Tab. 3: Obere Tafel: Korrelationsmatrix (DE3) der Hauptelemente auf Basis der mittels μ XRF gemessenen Intensitäten. Kursiv gedruckte Werte sind auf dem Niveau von 0,05 nicht signifikant. Untere Tafel: Faktorladungen der Hauptkomponentenanalyse, inklusive der Eigenwerte (Eigen.) und der jeweiligen beschriebenen Varianz (dVar.). Fett gedruckte Werte (\geq [0.4]) haben einen hohen Einfluß auf die einzelnen Hauptkomponenten.

	Mg	AI	Si	Р	к	Ca	ті	Mn	Fe
Mg	1	0.88	0.10	0.43	0.77	-0.48	0.56	0.29	0.58
AI	0.88	1	0.52	0.38	0.80	-0.24	0.28	0.05	0.29
Si	0.10	0.52	1	-0.15	0.40	0.40	-0.40	-0.45	-0.42
Р	0.43	0.38	-0.15	1	0.11	-0.21	0.25	0.40	0.35
к	0.77	0.80	0.40	0.11	1	-0.27	0.20	0.00	0.28
Ca	-0.48	-0.24	0.40	-0.21	-0.27	1	-0.72	-0.54	-0.82
Ті	0.56	0.28	-0.40	0.25	0.20	-0.72	1	0.59	0.88
Mn	0.29	0.05	-0.45	0.40	0.00	-0.54	0.59	1	0.79
Fe	0.58	0.29	-0.42	0.35	0.28	-0.82	0.88	0.79	1

	PC 1	PC 2	PC 3	PC 4	PC 5	PC 6	PC 7	PC 8	PC 9
Eigen	4.27	2.55	0.93	0.45	0.33	0.29	0.12	0.04	0.01
dVar.	53.3%	22.4%	9.2%	5.2%	4.6%	3.8%	1.0%	0.6%	0.1%
% Cumul.	53.3%	75.7%	84.9%	90.1%	94.7%	98.4%	99.4%	99.9%	100.0%
Mg	0.21	0.34	-0.10	0.11	-0.15	0.20	-0.60	-0.25	0.59
AI	0.10	0.39	-0.13	0.01	-0.09	-0.09	-0.44	-0.10	-0.77
Si	-0.22	0.54	0.00	-0.48	0.08	-0.59	0.14	0.01	0.22
Р	0.17	0.06	-0.81	0.38	-0.13	-0.24	0.28	0.09	0.06
К	0.14	0.63	0.12	0.14	0.21	0.53	0.39	0.27	-0.02
Ca	-0.48	0.00	-0.38	-0.44	-0.40	0.49	0.10	-0.17	-0.05
Ti	0.49	-0.02	0.18	-0.26	-0.70	-0.06	0.08	0.40	0.01
Mn	0.36	-0.19	-0.35	-0.53	0.50	0.16	-0.22	0.32	0.01
Fe	0.50	-0.02	0.02	-0.21	0.05	0.02	0.36	-0.75	-0.07

3.3 From geochemical stratification towards a dust signal

PCA is applied to reduce the amount of variables and to achieve an integrated signal which corresponds better to the lithology of the sediment than the single element stratications. In both cores the first principle component factor (PC 1) illustrated the change from facies containing glacial and aeolian sediment to those which are characterized by containing organic gyttja (Fig. 6). The discrimination into these two lithological groups gives reliable results for SM3 and DE3, which is also indicated by the described variances of the PC 1. This factor accounts for 53.3% in DE3 and for 56.9% in SM3, respectively (Tab. 3, 4). However, both cores show different factor loadings for the first variable. In core SM3 the main geogenic elements Si, Ti, and K represent those samples which are derived from aeolian sediment. In SM3 Calcium is an additional element characteristic of loess gyttja, and is also the most prominent element in the core DE3. The interpretation of the second PCA variable (PC 2) is more difficult, because of even larger differences between the single cores. In SM3 the PC 2 includes variables with high factor loadings on typical limnogenic and

redox-sensitive elements like Mn, Fe, and P. High values of P are provided by oxic decay of organic matter. Fe and Mn can be used to indicate the redox conditions of the lake water (ENGSTROM $\mathring{\sigma}$ WRIGHT 1984). When the lake is under anoxic conditions, Fe and Mn are present in soluble form and are easily released from the sediment. In DE3 the geogenic elements are the major contributors on PC 2. These elements are common constituents of primary minerals. Thus PC 2, with high factor loadings on geogenic elements, reflects the intensity of chemical weathering and leaching in the watershed and shows positive values from 35 m depth down core in DE3.

Summarizing these results one can find that analyses of the first principal component separates glacial and aeolian dominated sediment from interglacial/interstadial organicrich matter. However, in each core the elements with highest loadings differ in the respective factor. In SM3 the geogenic elements are those which characterize the first PC. In the core DE3 the geogenic elements describe the second PC, whereas Ca and Si are the most important elements for classification of the glacial sediment in PC 1. On the other hand redox-sensitive elements like Mn and P have high loadings in PC 2 for core SM3 and are an indicator for

Tab. 4: Upper panel: Correlation matrix of main elements (core SM3 measured intensities by μ XRF). Italic printed values are not significant at the 0.05 level. Lower panel: Factor loadings after PCA calculation of main elements (core SM3 measured intensities by μ XRF), including Eigenwert (Eigen.), the described variance (dVar.) and the cumulative sum of dVar (% Cumul). Bold printed values ($\geq |0.4|$) have a major influence on the single principle components.

Tab. 4: Obere Tafel: Korrelationsmatrix (SM3) der Hauptelemente auf Basis der mittels μ XRF gemessenen Intensitäten. Kursiv gedruckte Werte sind auf dem Niveau von 0,05 nicht signifikant. Untere Tafel: Faktorladungen der Hauptkomponentenanalyse, inklusive der Eigenwerte (Eigen.) und der jeweiligen beschriebenen Varianz (dVar.). Fett gedruckte Werte (\geq [0.4]) haben einen hohen Einfluß auf die einzelnen Hauptkomponenten.

	Mg	AI	Si	Р	к	Ca	Ті	Mn	Fe
Mg	1	0.88	0.59	0.47	0.75	0.42	0.85	0.08	0.54
AI	0.88	1	0.67	0.52	0.81	0.40	0.89	0.02	0.48
Si	0.59	0.67	1	0.07	0.85	0.66	0.74	-0.35	0.50
Р	0.47	0.52	0.07	1	0.12	-0.16	0.30	0.60	0.38
к	0.75	0.81	0.85	0.12	1	0.69	0.94	-0.31	0.56
Ca	0.42	0.40	0.66	-0.16	0.69	1	0.61	-0.32	0.41
Ті	0.85	0.89	0.74	0.30	0.94	0.61	1	-0.14	0.57
Mn	0.08	0.02	-0.35	0.60	-0.31	-0.32	-0.14	1	0.25
Fe	0.54	0.48	0.50	0.38	0.56	0.41	0.57	0.25	1

	PC 1	PC 2	PC 3	PC 4	PC 5	PC 6	PC 7	PC 8	PC 9
Eigen	5.144	2.0437	0.7322	0.3646	0.2889	0.2024	0.1461	0.0553	0.0227
dVar.	56.9%	23.8%	7.5%	3.8%	3.2%	2.2%	1.7%	0.6%	0.3%
% Cumul.	56.9%	80.8%	88.2%	92.0%	95.2%	97.5%	99.1%	99.7%	100.0%
Mg	0.39	-0.18	-0.19	-0.31	-0.33	0.13	0.72	-0.20	0.08
AI	0.42	-0.16	-0.34	-0.10	0.01	-0.18	-0.15	0.78	0.05
Si	0.36	0.20	0.06	0.45	0.33	-0.60	0.30	-0.13	-0.21
Р	0.16	-0.60	-0.15	0.18	0.62	0.36	-0.03	-0.19	0.06
К	0.42	0.15	-0.03	0.08	-0.13	-0.08	-0.39	-0.34	0.71
Ca	0.30	0.32	0.56	-0.50	0.44	0.19	0.05	0.13	0.02
Ті	0.43	0.01	-0.11	-0.14	-0.20	0.11	-0.45	-0.31	-0.66
Mn	-0.06	-0.63	0.44	-0.29	-0.12	-0.53	-0.12	-0.09	0.02
Fe	0.25	-0.16	0.55	0.55	-0.36	0.34	0.05	0.25	-0.03



Fig. 6: The first two principal components PC1 and PC2 of each core are shown versus the depth (see tables 3b, 4b for associated factor loadings). (a, b) PC1 isolates a signal of glacial/aeolian transported sediment (shaded in blue). Aeolian sediments have negative values in the core DE3 and positive ones in the core SM3, respectively (*For a better visualization the x-axis of (b) is inverted). (c, d) PC2 in both cores has in common that they have negative values during warm/humid climate conditions (red color). However, in core SM3 PC2 is characterized by highly redox-sensitive elements (see text). PC2 of core DE3 depends strongly on elements which indicate enhanced chemical weathering.

Abb. 6: Darstellung der beiden ersten Hauptkomponenten (PC1, PC2) gegen die Tiefe des jeweiligen Sedimentkerns (s. Tab. 3b, 4b für dazugehörige Faktorladungen). (a, b) Äolischer Staub (blaue Füllung) ist im Kern DE3 durch negative PC1-Werte gekennzeichnet, im Kern SM3 durch positive Werte. Zur Verdeutlichung des staubhaltigen Sediments ist die x-Achse der PC1 des Kerns SM3 invers dargestellt. (c, d) PC 2 ist in beiden Kernen unterschiedlich zu interpretieren deutet aber auf klimatisch warm/humiden Einfluss hin (rote Färbung). Im Kern SM3 ist PC2 durch redox-sensitive Elemente charakterisiert (s. Text). Im Kern DE3 hängt PC2 stark vom Vorkommen von Elementen ab, die auf erhöhte chemische Verwitterung hindeuten. the redox conditions at the lake bottom with contemporaneous high amount of organic matter. The other factors do not influence aeolian sediment input into the maar lake and account only for a lower percentage of the total variation. In addition, the result of the scree test suggests using the first two factors, only. Consequently, they are of minor interpretative use for this study and not further considered.

4 Discussion

For the derivation of paleoclimate or economic conditions from geochemical data two assumptions must hold true (BOYLE, 2001): (1) At any one time the sediment concentration must be proportional to the external element loading and (2) the element concentrations in the sediment must not change after burial. The latter is in discussion especially as regards the role of Mn and Fe reduction in potential remobilization of trace elements. BOYLE (2001) has shown that post depositional alteration and migration is likely, but that this is only significant at extremely low sediment mass accumulation rates and is not likely for high glacial sediment yields. Thus we suggest that the elicitation of a dust signal within maar lake sediments is possible by application of inorganic geochemistry. However, a wider knowledge about the lake's regional setting (location, morphology, presence of inflows, etc) is necessary. Thus, additional information from other proxies is required. This is shown in this study, since all cores show different inorganic geochemical reactions to environmental changes dependent on the regional setting. An example is the stratification of the geogenic elements in the core SM3: All of these elements have a major influence on PC 1, but only the typical "loess-elements" Si, Ca, and K have their highest values during glacial conditions.

The results show that there are three different groups of main elements which characterize the sediment. (i) Typical geogenic elements are magnesium (low differentiated volcanic products), potassium (low differentiated volcanic products), titanium and aluminum (lithics, feldspars). In oligotrophic lakes silica, aluminum as well as titanium and potassium represent the clastic input. This is clearly shown for all the cores. The alkali metal K and the alkaline earth metals Ca and Mg are major bedrock constituents. High amounts of these elements represent reduced soil stability and increased erosion. But Ca and Mg might have also significant concentrations in authigenic carbonate and skeletal carbonates from invertebrates. The supply of conservative cations K (and also Na) increases with periods of rapid chemical erosion, similar to Ti. These elements are thus enriched during the Holocene and MIS-3 and only minor during MIS-2, when climate was more arid. (ii) Sulphur and phosphorous belong are limnogenic and redox-sensitive elements and thus the precipitation and release processes are complex. The deposition and conservation of the elements Fe, Mn, S, and P is strongly controlled by pH and redox conditions at the lake bottom (ENGSTROM & WRIGHT 1984). Since maar lakes usually have anoxic bottom water reduced Fe(II) precipitates in combination with S, forming the mineral pyrite (FeS₂). Blue colored vivianite crystals, a ferrous phosphate, occur with increased burial of undecomposed organics. High rates of organic matter deposition



Fig. 7: μ XRF geochemistry stacks versus the time, (a) NorthGRIP d180 record, (b) the ELSA grayscale stack, (c) the ELSA dust stack, and major litho zones of the last 60 kyrs. Ca (d) as well as the PCA stack (e) clearly shows the influence of aeolian sediment (* The SM3 part of the PC 1-stack is inverse; see Fig. 6). Heinrich events after HEMMING (2004).

Abb. 7: µXRF-Geochemie-Stacks gegen das Alter: (a) NorthGRIP d18O-Rekord, (b) der ELSA Graustufen-Stack, (c) der ELSA Staub-Stack und die wichtigsten Lithozonen der Bohrkerne. Die Ca-Werte (d) sowie der PC1-Stack (e) zeigen deutlich den Einfluss des äolischen Sediments (*der SM3-Abschnitt des PC1-Stacks ist invers aufgetragen; siehe Abb. 6). Marine Heinrich-Ereignisse nach НЕММІNG (2004).

occur only during interstadials and interglacials. A second prerequisite for the formation of vivianite is the presence of orthophosphates which are provided by oxic decay. Thus, the occurrence of vivianite in organic rich sediment sections is not a proxy for paleoproductivity (FAGEL et al., 2005). It is an indicator of at least temporary deep circulation at high organic production (VOIGT et al., 2008) and clearly precedes the formation of sulphides in the course of eutrophication (GÄCHTER & MÜLLER, 2003). The presence of Pyrite shows on the other hand the level of aeration of the maar deep water and the occurrence of a monimolimnion, with anoxic water conditions for at least a period of several mixing cycles. Vivianite is found in the cores with a high content of Phosphorous in the DE3 47.7-50 m and at 70.7-72.7 m depth, in particular. (iii) The DE3 at third class is a mixed type: Silicon represents minerogenous SiO₂ in the clastic fraction as well as amorphous and autogenic SiO₂ production, mainly by the silicon shell of diatoms. Calcium is a major constituent of aeolian dust sediments (with provenance of the Devonian or Triassic limestone). Further reasons for high contents of Ca are the precipitation of CaCO₃ caused by detraction of CO₂ in the lake water, and biogenic calcium precipitation e.g. in the lime shell of ostracods. Authigenic carbonates were not observed during glacial periods in the core DE3 (DIETRICH & SEELOS 2010) and we suggest this for the other glacial sequences, too. During times of dominant physical weathering CaCO₃ grains will be transported to the lake and most of them should quickly reach the lake bottom without significant in-lake chemical reaction (KOINIG et al. 2003). Thus, we suggest that the content of calcium has not changed during settlement at glacial conditions and can be interpreted as a major contributor to the measured dust signal. However, a stack of the elemental stratification of different cores does not seem to be appropriate. The influence of the catchment, different biogeochemical processes in the lake water and bottom makes each maar lake unique, which is reflected in the distribution of the elements.

The focus of this study is on the detection of aeolian sediments by μ XRF scanning and PCA. It is shown that the methodology gives reliable results within acceptable error margins but still needs additional information, at least from facies and micro-facies analysis. However, in this study

time slices including major climate changes of the last glacial were used to evaluate this approach and are already discussed in early ELSA publications: e.g. the late Weichselian glacial inception and Heinrich events during MIS-3 and 2 (DIETRICH & SEELOS 2010), transition I, the Younger Dryas and internal variability of the Holocene (SEELOS et al. 2009). A major finding of this study is that the previously used approach demonstrates weaknesses in comparison to high resolution grain size analysis which was used in the mentioned publications by applying the RADIUS technique (SEELOS & SIROCKO 2005). For example, the variability of MIS-2 dust sources could only be shown by a combination of mineralogical/geochemical analysis and grain size sorting parameters (DIETRICH & SEELOS 2010). However, the transition into the glacial climate state is shown within an abrupt trend towards negative values of PC 1 in the core DE3. This signal is highly correlated to the dust detection for the same core published as ELSA dust stack (Fig. 8). The timing of increased dust transport towards the Eifel maar lakes correspond with the global decrease of sea-level 35 kyrs ago (CLARK et al. 2009). An increased east wind frequency for Western Europe was suggested by DIETRICH & SEELOS (2010) for the time slice between 33 and 24 kyrs BP. These layers are enriched in aeolian transported carbonate grains in the same sediment core DE3. Since, Ca has a major influence of DE3's PC1, these layers are also reflected in PC 1 (Fig. 7d, e). In addition a severe cold phase which corresponds to the strong Heinrich event H4 (HEMMING 2004) is detected by the PCA approach. Another major dust event correlates with H5 (SEELOS et al. 2009) which is reworked by debris in the core DE3 at 65 m depth, and thus could not be shown by the applied approach.

The younger sediment sequence shows in core SM3 the major climate change from the last glacial to interglacial conditions of the Holocene. Fast environmental changes from the last glacial to the interglacial including the YD are also documented in this core. However, the tephra of the lake Laach eruption shows negative factor scores, too, and is wrongly classified as glacial and aeolian sediment. PC 1 shows a high variability during the Holocene, which is also shown in the lithology and in the grayscale values of the core (Fig. 7). The strong shifts in the aeolian sedi-

ment proxy in the Holocene (especially the Atlanticum) might also be influenced by changing lake levels during the Holocene. Correlations between the North Atlantic IRD events defined by BOND et al. (1993) and lake level changes in mid-European lakes have been pointed out (MAGNY 2004, 1999; FRIEDRICH et al. 1999), supporting the hypothesis that changes in solar activity have been an important forcing factor of Holocene climate (RENSSEN & ISARIN 2001). A pronounced Ca peak corresponds to the strong cold spell of the 8.2-ka-event. This event is only minor correlated with an enhanced dust input in the Eifel area (SEELOS et al. 2009). Here, we suggest that a strong ventilation of the bottom water has led to the precipitation of carbonate minerals. However, PRASAD et al. (2009) suggested temperature to be the primary control on calcite formation. These authors focused on seasonal carbonate precipitates in the sublayers of single warves and found distinct minima of calcite during the 8.2-ka-event which were used as an proxy for cold summer temperatures. In deed, the calcium maximum in SM3 is not continuously on a high level when looking at high resolution μ XRF scans, but also shows several phases with minimum values (not shown). The major impact of the 160 year 8.2-ka-event on the environment of Central Europe is also shown in many other records e.g. in reduced tree-ring widths from Central German oaks (FRIEDRICH et al. 1999). However, in the Holocene the deposition of aeolian dust plays only a minor role. The elements Al, K, Ti have the highest factor loadings in PC 1 of SM3 and reflect soil destabilization which goes along with erosion and enhanced dust transport. Until the Boreal the environment is characterized by more open landscapes and high physical weathering. These conditions change within the Atlanticum when conditions changed to a warmer and more humid climate. Hence, the soils became more stabilized and the slopes were covered by dense forests. Human activities in the catchment are traced back to ca. 15.5 kyrs BP, long before late glacial warming ca. 14.7 kyrs BP (STREET et al. 2006), but became most pronounced around 3,700 BP (SIROCKO, ed. 2009). Deforestation, grazing, and agriculture caused an increase in erosion, physical weathering and thus enhanced deflation of dust. Since then the dust signal in the lake Schalkenmehrener Maar is mainly anthropogenically altered.

5 Conclusion

This study investigates whether the detection of aeolian sediment by μ XRF scanning of resin impregnated sediment samples with a subsequent PCA approach give reliable results. Changes of aeolian supply over time are suggested in the inorganic geochemistry data, especially in Ca, Si, and the first new calculated PCA factor (PC 1) which is mainly influenced by Ca, Si, and Ti. It is demonstrated that this approach works well since for the interpretation further knowledge about the lithology and environmental background is available. The signal of geochemical aeolian proxies might be influenced by autochthonous and allochthounous accumulation changes and without further knowledge and necessary inter proxy comparison the results might be inconsistent or difficult to interpret. Thus, the detection of dust using high resolution grain size analysis is advantageous, because this

approach recognizes different transport mechanisms of the sediment and is so far the only method which is able to quantify the content of aeolian dust in sediment. A μ XRF is a fast and inexpensive scanning method of the elemental stratification within sediment cores. In addition, the quantification of the main elements via standard less fundamental parameter method gives significant results in comparison to the reference samples, which are measured by WD-XRF. Thus, a calibration of the μ XRF results is reliable, if this evaluation is based on at least a reference sample for each litho zone.

Cores SM3 and DE3 show both fully glacial sediment and warm and wet climate conditions and the major dust deposition events and phases could be detected by µXRF geochemistry. These include the largest Heinrich event H4 during MIS-3, the onset of dust increase coupled to the growing of the Scandinavian Ice sheet, later on the whole MIS-2 including LGM, the YD as well as enhanced dust supply generated by human activities since the Subboreal. In the applied PCA approach PC 1 discriminates in both lake sediment cores aeolian from non aeolian material. Volcanic ash and tephra is mostly classified as aeolian sediment because of the major influence of geogenic elements (Si, Al, Ti, K and mostly Ca) to PC 1. Ca has a major influence on the dust signal in both cores and a combination of Ca with the grayscale values gives the best signal of aeolian sediment. The occurrence of vivianite coincides with enriched amounts of P and Fe in organic rich gyttja and is thus suggested as a proxy for low sedimentation rates during interstadial climate conditions. In summary, µXRF scanning of bulk geochemistry has shown to be a sensitive indicator of both changes in the lake and in its (aeolian) catchment, if it is used together with other proxy data such as micro facies or grain size analysis.

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