



Supplement of

Ideas and perspectives: hydrothermally driven redistribution and sequestration of early Archaean biomass – the "hydrothermal pump hypothesis"

Jan-Peter Duda et al.

Correspondence to: Jan-Peter Duda (jan-peter.duda@geo.uni-goettingen.de)

The copyright of individual parts of the supplement might differ from the CC BY 4.0 License.



Fig. S1. Hydrothermal chert veins of the ca. 3.5 Ga Dresser Formation (Pilbara Craton, Western Australia). (a) Hydrothermal chert veins of the Dresser Formation (ridges, see arrows) forming large-scale networks in their host basalts. (b) Hydrothermally altered footwall basalts exhibiting pillow structures (arrows); hammer for scale (red circle). (c, d) Hydrothermal chert veins of the Dresser Formation penetrating komatiitic footwall basalts in a recent cut wall of the

abandoned Dresser Mine (persons for scale). The analysed hydrothermal chert vein occurs adjacent to the one shown in (d).

5



Fig. S2. Total ion current chromatograms. Low-temperature (**a**) and high-temperature (**b**) HyPy products of the analytical blank (combusted sea sand) obtained prior to HyPy of the Dresser kerogen. Low-temperature (**c**) and high-temperature (**d**) HyPy products of the Dresser kerogen. Compounds detected in (**a**–**c**) represent background contamination and/or artefacts. Note that high-temperature HyPy of the Dresser kerogen yielded significantly higher amounts of products with a distinctly

5 different distribution pattern. Black dots: *n*-alkanes (numbers refer to carbon chain-lengths); triangle: phthalic acid; N: naphthalene; MN: methylnaphthalenes; BiPh: 1,1'-biphenyl; DMN: dimethylnaphthalenes; MAN: methylacenaphthenes; P: phenanthrene; crosses: siloxanes (GC column or septum bleeding); squares: phenols; S: sulphur.

Note: Percentage values given on the vertical axes of chromatograms (**a**–**c**) relate peak intensities to chromatogram (**d**) (HyPy Dresser kerogen, 330–520°C).



Fig. S3. Partial ion chromatograms selective for alkanes (m/z 85). Low-temperature (**a**) and high-temperature (**b**) HyPy products of the analytical blank (combusted sea sand) obtained prior to HyPy of the Dresser kerogen. Low-temperature (**c**) and high-temperature (**d**) HyPy products of the Dresser kerogen. High-temperature HyPy produced the highest yields of *n*-alkanes and minor clusters of isomeric monomethylalkanes (diamonds in **d**). The *n*-alkanes in the high-temperature

5 pyrolysate of the Dresser kerogen (d) furthermore exhibit a distinct distribution different to those observed in (a-c). All compounds detected in (a-c) are considered to represent background contamination. Black dots: *n*-alkanes (numbers refer to carbon chain-lengths).

Note: Percentage values given on the vertical axes of chromatograms (a-c) relate peak intensities to chromatogram (d) (HyPy Dresser kerogen, 330–520°C).



Retention time

Fig. S4. Ion chromatograms selective for aromatic hydrocarbons (m/z 128, 142, 154, 156, 168, 178). Low-temperature (**a**) and high-temperature (**b**) HyPy products of analytical blank (combusted sea sand) obtained prior to HyPy of the Dresser kerogen. Low-temperature (**c**) and high-temperature (**d**) HyPy products of the Dresser kerogen. Note that high-temperature HyPy of the Dresser kerogen yielded a variety of aromatic hydrocarbons, which are orders of magnitudes lower or absent in

5 all other pyrolysates. Black dots: *n*-alkanes (numbers refer to carbon chain-lengths); N: naphthalene; MN: methylnaphthalenes; BiPh: 1,1'-biphenyl; DMN: dimethylnaphthalenes; AN: acenaphthene; MBiPh: methylbiphenyls; DBF: dibenzofuran; MAN: methylacenaphthenes; P: phenanthrene; crosses: siloxanes (GC column or septum bleeding); S: elemental sulphur (likely derived from the sulfidic catalyst).

Note: Percentage values given on the vertical axes of chromatograms (a-c) relate peak intensities to chromatogram (d)10 (HyPy Dresser kerogen, 330–520°C).

7



Fig. S5. Partial ion chromatograms selective for (dimethyl-, methyl-)naphthalenes (m/z, 128, 142, 156). Low-temperature (a) and high-temperature (b) HyPy products of the analytical blank (combusted sea sand) obtained prior to HyPy of the Dresser kerogen. Low-temperature (c) and high-temperature (d) HyPy products of the Dresser kerogen. High-temperature HyPy of the Dresser kerogen yielded naphthalene (N), methylnaphthalenes (MN), dimethylnaphthalenes (DMN) and acenaphthene

5

(AN), which are orders of magnitudes lower or absent in all other pyrolysates. Black dots: n-alkanes (numbers refer to carbon chain-lengths); S: elemental sulphur (likely derived from the sulfidic catalyst).

Note: Percentage values given on the vertical axes of chromatograms (a-c) relate peak intensities to chromatogram (d)(HyPy Dresser kerogen, 330–520°C).



Fig. S6. Partial ion chromatograms selective for (methyl-)phenanthrenes (m/z 178, 192). Low-temperature (**a**) and high-temperature (**b**) HyPy products of the analytical blank (combusted sea sand) obtained prior to HyPy of the Dresser kerogen. Low-temperature (**c**) and high-temperature (**d**) HyPy products of the Dresser kerogen. Phenanthrene (P) and traces of methylphenanthrenes (MP) were only present in the high-temperature HyPy pyrolysate of the Dresser kerogen. Squares: phenols; S: elemental sulphur (likely derived from the sulfidic catalyst).

5

Note: Percentage values given on the vertical axes of chromatograms (**b**–**d**) relate peak intensities to chromatogram (**a**) (HyPy blank, \sim 21–330°C).



Fig. S7. Ion chromatograms selective for branched alkanes with quaternary carbon centres (BAQCs; m/z 127). Low-temperature (**a**) and high-temperature (**b**) HyPy products of the analytical blank (combusted sea sand) obtained prior to HyPy of the Dresser kerogen. Low-temperature (**c**) and high-temperature (**d**) HyPy products of the Dresser kerogen. Compounds detected in (**a**–**c**) represent background contamination and/or artefacts. Note the absence of BAQCs in all

5 pyrolysates. Black dots: *n*-alkanes (numbers refer to carbon chain-lengths); diamonds: monomethylalkanes; N: naphthalene; BiPh: 1,1'-biphenyl; crosses: siloxanes (GC column or septum bleeding); S: elemental sulphur (likely derived from the sulfidic catalyst).

Note: Percentage values given on the vertical axes of chromatograms (a-c) relate peak intensities to chromatogram (d) (HyPy Dresser kerogen, 330–520°C).



Fig. S8. Ion chromatograms selective for hopanes (m/z 191). Low-temperature (**a**) and high-temperature (**b**) HyPy products of the analytical blank (combusted sea sand) obtained prior to HyPy of the Dresser kerogen. Low-temperature (**c**) and high-temperature (**d**) HyPy products of the Dresser kerogen. All compounds in (**a**–**d**) represent background contamination and/or artefacts. Note the absence of hopanes in all pyrolysates. Crosses: siloxanes (GC column or septum bleeding); squares:

5 phenols.

Note: Percentage values given on the vertical axes of chromatograms (**c**–**d**) relate peak intensities to chromatogram (**a**) (HyPy blank, \sim 21–330°C).



Fig. S9. Ion chromatograms selective for steranes (m/z 217). Low-temperature (**a**) and high-temperature (**b**) HyPy chromatograms of the analytical blank (combusted sea sand) obtained prior to HyPy of the Dresser kerogen. Low-temperature (**c**) and high-temperature (**d**) HyPy products of the Dresser kerogen. Note the absence of steranes in all chromatograms.

5 Note: Percentage values given on the vertical axes of chromatograms (**a**–**c**) relate peak intensities to chromatogram (**d**) (HyPy Dresser kerogen, 330–520°C).



Fig. S10. Stable carbon isotope values (δ^{13} C) of *n*-alkanes released upon high-temperature HyPy and the total organic carbon (TOC). The isotopic similarity indicates that the *n*-alkanes (black dots) were generated from the kerogen (TOC, red dot). Vertical lines: Standard deviations of δ^{13} C values; dotted horizontal line: mean δ^{13} C value of *n*-alkanes (-31.4 %); shaded

area: standard deviation of mean δ^{13} C value of *n*-alkanes (±1.2 ‰).

5